

Note

A ^{13}C -n.m.r. analysis of linkages in lichen polysaccharides: an approach to chemical taxonomy of lichens

ITSURO YOKOTA*†, SHOJI SHIBATA*††, AND HAZIME SAITÔ**

Faculty of Pharmaceutical Sciences, University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113 (Japan), and Biophysics Division, National Cancer Center Research Institute, Tsukiji 5-chome, Chuo-ku, Tokyo 104 (Japan)***

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It has been found that various kinds of water-soluble polysaccharides are contained in lichens in fairly high proportions; of these polysaccharides, lichenan¹, isolichenan¹, and pustulan² had been extensively studied to elucidate their chemical structures. The polysaccharides thus far known in lichens are the homo-D-glucans (1→3)-(1→4)-β-D-glucan (lichenan), (1→3)-(1→4)-α-D-glucan [isolichenan¹, PC-3 (ref. 3), EP-3 (ref. 4), and everniin⁵], (1→6)-β-D-glucan [pustulan and GE-3 (ref. 6)] and (1→3)-(1→4)-(1→6)-α-D-glucan (acroscyphan⁴).

In contrast to the lichen metabolites of lower molecular weight, the distribution of which is mostly species-specific⁷, the lichen polysaccharides are rather characteristic of larger taxonomical groups^{7, 8}. Hence, a wide survey of lichen polysaccharides was undertaken. It has also been demonstrated that some polysaccharides of lichens and higher fungi show host-mediated, antitumor activity against transplanted Sarcoma 180 in mice^{4, 6, 9-11}.

We now describe the isolation and ^{13}C -n.m.r. analysis of some lichen homo-D-glucans and their relative proportions in the lichens: the lichenan and isolichenan from *Cetraria richardsonii*, SJ-2-I from *Stereocaulon japonicum*¹² Th. Fr., PA-2 from *Pilophoron acicularis* (Svj.) Nyl., PC-3 from *Parmeria caperata* (L.) Ach., acroscyphan from *Acroscyphus sphaerophoroides* Lev., and SG-3 from *Sphaerophorus globosus* (Huds.) Vain.

EXPERIMENTAL

General. — Sugar analysis was performed with a JEOL JLC-6AH liquid-

*Present address: Nippon Chemiphar Co. Ltd., 1-22 Hikokawato, Misato-shi, Saitama-ken, 341, Japan.

††Present address: Meiji College of Pharmacy, 1-35-23 Nozawa, Setagaya-ku, Tokyo, 154, Japan.

chromatographic autoanalyzer. Infrared spectra were recorded with a Jasco DS 402-G spectrophotometer. ^{13}C -N.m.r. spectra were recorded with a JEOL PFT-100/EC-100 spectrometer operating at 25.03 MHz, in the pulsed, Fourier-transform mode. Free-induction decays were accumulated with a 45° pulse and a repetition time of 1.2 sec. All spectra were recorded using 4K data-points and spectral width of 4 kHz. ^{13}C chemical shifts were expressed in p.p.m. downfield from external tetramethylsilane. All samples (80 mg/ml; pH 14) were contained in tubes (10 mm, o.d.).

Isolation and purification of SG-3 and PA-2. — SG-3. Thalli (42.6 g) of the lichen *Sphaerophorus globosus* were successively extracted with Me_2CO and 80% EtOH in order to remove soluble components. The residual thalli were further extracted with dist. H_2O on a boiling-water bath. The hot extract was concentrated *in vacuo*, and the concentrate was poured into EtOH (3 vol.) to afford a precipitate which was collected by centrifugation and dried, to give a pale-grey, water-soluble substance (polysaccharide fraction SG-1; yield 7.7 g, 18.1%). SG-1 was further separated into a cold-water-soluble fraction, SG-2 (yield 6.6%), and an insoluble fraction, SG-3 (1.9%), by the procedure of freezing and thawing. By determination with a sugar analyzer of the products of acid hydrolysis, it was proved that SG-2 contained mannose (major), galactose, and glucose, whereas SG-3 contained glucose only. Fraction SG-3 showed $\nu_{\text{max}}^{\text{KBr}}$ 844 cm^{-1} .

PA-2. — Thalli (292 g) of the lichen *Pilophoron acicularis* were extracted by the procedure just described. The resulting, pale-brown, water-soluble substance, PA-1 (yield 28.6 g, 9.8%), was separated into a cold-water-soluble fraction, PA-2 (yield 5.0%), and an insoluble fraction, PA-3 (yield 0.05%). Fractions PA-1, PA-2, and PA-3 contained glucose only, and showed $\nu_{\text{max}}^{\text{KBr}}$ 845, 849, and 850 cm^{-1} , respectively. Fraction PA-2 gave a single peak in gel filtration on Sephadex G-200.

Preparation of lichenan, isolichenan, PC-3, SJ-2-I, and acroscyphan. — The isolation procedures were essentially the same as those already described^{3,4,12}, and the physical and chemical data were identical with the values previously reported.

RESULTS AND DISCUSSION

Fig. 1 shows the ^{13}C -n.m.r. spectra of some lichen polysaccharides. The assignment of the ^{13}C signals of PC-3 [(1 \rightarrow 3)-(1 \rightarrow 4)- α -D-glucan containing each component in 1:1 ratio] was achieved from consideration of the relative peak-intensities (total 12), and the ^{13}C chemical shifts of (1 \rightarrow 4)- α -D-glucans and (1 \rightarrow 3)- α -D-glucans^{13,14}. The assignment was confirmed by examination of the ^{13}C -n.m.r. spectra of other D-glucans, having different relative-intensities (see Table I). In a similar way, the ^{13}C -n.m.r. signals of lichenan were assigned on the basis of those given by (1 \rightarrow 3)- β -D-glucans and cellobiose¹⁴.

The ratio of (1 \rightarrow 3)- to (1 \rightarrow 4)-linkages was determined by comparing the integrated peak-intensity of C-3 [(1 \rightarrow 3)-linkage] with that of C-4 [(1 \rightarrow 4)-linkage] for both α - and β -D-linked D-glucans, as no overlaps of peaks appear in this region. The relative proportion of α -D-(1 \rightarrow 6)-linkages was estimated in the region of the

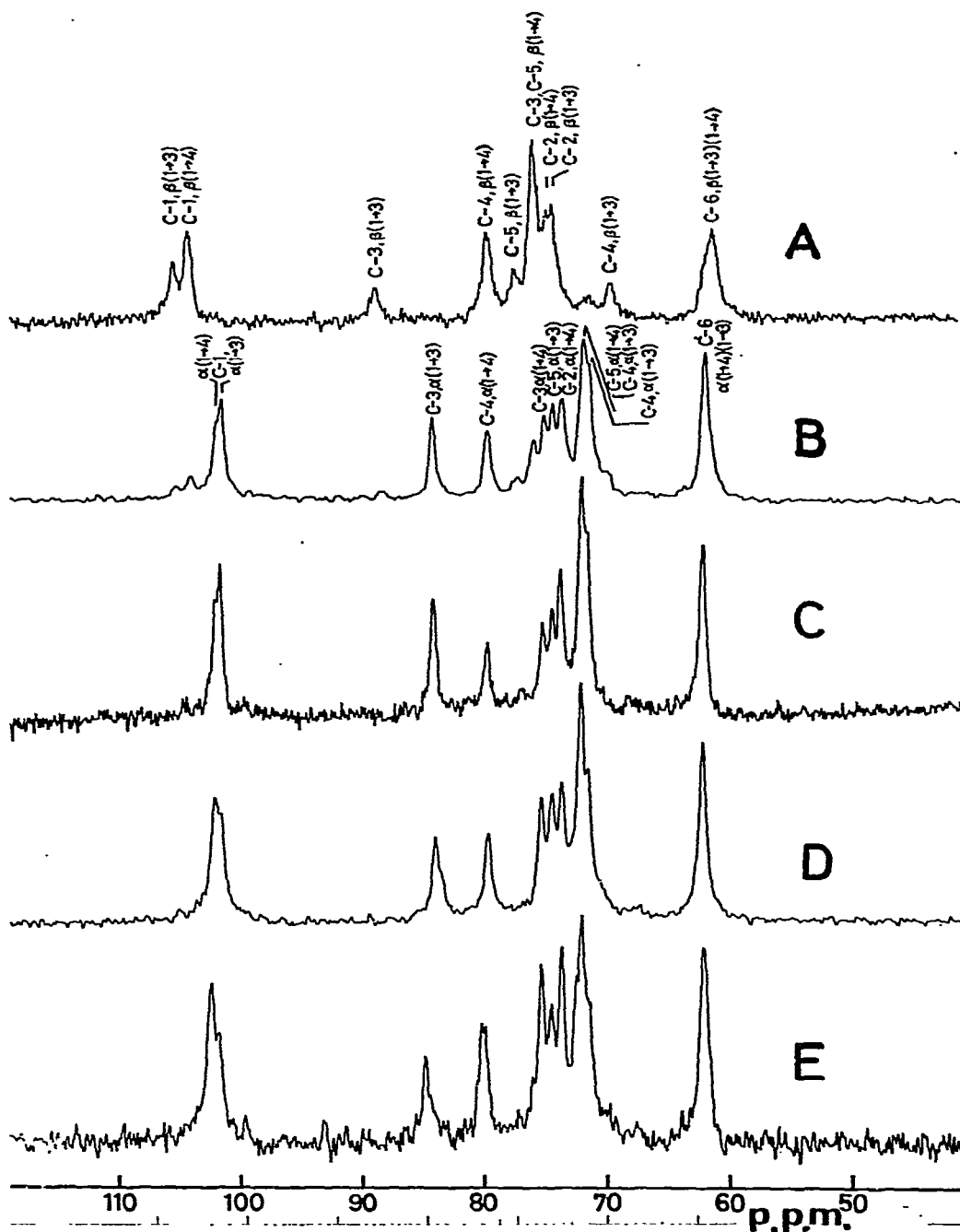


Fig. 1. ^{13}C -N.m.r. spectra of lichen polysaccharides in aqueous solution (80 mg/ml, pH 14). [A. Lichenan, 72,900 transients; B. Isolichenan, 75,120 transients; C. SJ-2-I, 13,000 transients; D. PC-3, 81,168 transients; and E. SG-3, 35,000 transients.]

TABLE I

¹³C-N.M.R. SIGNALS, AND THEIR ASSIGNMENTS, FOR LICHEN POLYSACCHARIDES^a

Lichen polysaccharides [α -(1 \rightarrow 3)-(1 \rightarrow 4)]					D-Glucans having single D-glucosidic linkages ^b				
PC-3	Isolichenan	SI-2-I	PA-2	SG-3	Acroscyphan	Assignment	α -(1 \rightarrow 4)	α -(1 \rightarrow 3)	α -(1 \rightarrow 6)
101.9(1) ^c	101.9	102.0	102.0	102.3	102.0	C-1 (1 \rightarrow 4)	102.9		
101.3(1)	101.6	101.5	101.8	101.7	101.5	C-1 (1 \rightarrow 3)		101.3	99.4 (C-1)
83.7(1)	84.3	84.1	84.7	84.7	83.5	C-3 (1 \rightarrow 3)		83.2	
79.3(1)	79.8	79.6	79.9	80.1	79.6	C-4 (1 \rightarrow 4)	80.6		
75.0(1)	75.1	75.1	75.1	75.2	75.1	C-3 (1 \rightarrow 4)	75.4		75.4 (C-3)
74.2(1)	74.4	74.3	74.4	74.3	74.4	C-5 (1 \rightarrow 3)		73.7	
73.3(1)	73.6	73.6	73.7	73.5	73.4	C-2 (1 \rightarrow 4)	73.8		73.1 (C-2)
71.8(2)	71.9	71.9	71.9	71.9	71.9	C-5 (1 \rightarrow 4)	72.6	72.2	71.8 (C-4)
						C-2 (1 \rightarrow 3)			
71.2(1)	71.4	71.5	71.5	71.4	71.4	C-4 (1 \rightarrow 3)		71.7	71.1 (C-5)
61.9(2)	62.0	62.0	62.0	62.0	61.9	C-6 (1 \rightarrow 3)	62.0	62.2	66.8 (C-6)
						C-6 (1 \rightarrow 4)			
(1:1) ^d	(3:2) ^d	(2:1) ^d	(2:1) ^d	(2:3) ^{d,e}	(2:3) ^{d,e}				
1:1 ^f	3:2 ^f	3:1 ^f	—	—	—				

TABLE 1 (continued)

<i>Lichen polysaccharide</i> [β -(1 \rightarrow 3)-(1 \rightarrow 4)]	<i>Assignment</i>	<i>D-Glucans having single glucosidic linkage^g</i>	
		β -(1 \rightarrow 3)	β -(1 \rightarrow 4)
105.6	C-1 (1 \rightarrow 3)	104.7	
104.5	C-1 (1 \rightarrow 4)		104.5
89.2	C-3 (1 \rightarrow 3)	88.0	
79.9	C-4 (1 \rightarrow 4)		71.4
76.2	C-5 (1 \rightarrow 3)	77.8	
	C-3,5 (1 \rightarrow 4)		78.2
75.0	C-2 (1 \rightarrow 4)		74.9
74.6	C-2 (1 \rightarrow 3)	74.9	
69.8	C-4 (1 \rightarrow 3)	69.9	
62.0	C-6 (1 \rightarrow 3)	62.5	
61.5	C-6 (1 \rightarrow 4)		62.1
(3:7) ^d			
3:7 ^f			

^aP.p.m. from Me₄Si. ^bData taken from ref. 13. ^cRelative peak-intensity. ^dRatio of the integrated peak-intensity of C-3 [(1 \rightarrow 3)-linkage] to that of C-4 [(1 \rightarrow 4)-linkage]. ^ePeak for α -(1 \rightarrow 6)-linkage less than 6%. ^fData by chemical analysis. ^gData for cellobiose, assignment based on ref. 14.

C-1 peak (99.4 p.p.m.). The composition thus obtained is in excellent agreement with the data given by chemical analysis (see Table I) for lichenan, isolichenan, and PC-3. The relative proportions of the peaks other than those for C-3 [(1→3)-linkage] and C-4 [(1→4)-linkage] are also consistent with the values for the polysaccharides shown in Fig. 1.

From Table I, it may be seen that the ratio of the α -D-(1→3)- to α -D-(1→4)-glucosidic linkages in acroscyphan, a polysaccharide from *Acroscyphus sphaerophoroides*, is 2:3, and that it contains, at most, 6% of α -D-(1→6)-linkages; the chemical evidence provided earlier had revealed the presence of α -D-(1→3)-, -(1→4)-, and -(1→6)-linkages in acroscyphan, but did not give their ratios. Polysaccharide SG-3 from *Sphaerophorus globosus* has been indicated by the ^{13}C analysis to have α -D-(1→3)-(1→4)-linkages in the ratio of 2:3, with not more than 6% of α -D-(1→6)-linkages. The results of the ^{13}C -n.m.r. study and the chemical evidence now strongly suggest that SG-3 is identical with acroscyphan; this would be reasonable, as the two lichens, *Acroscyphus sphaerophoroides* and *Sphaerophorus globosus*, belong to the same family, the *Sphaerophoraceae*. Furthermore, it is found that the ^{13}C -n.m.r. data for PA-2 from *Pilophoron acicularis* are identical with those for SJ-2-I, from *Stereocaulon japonicum*, in which the ratio of α -(1→3)- to α -(1→4)-linkages was determined as 3:1 from chemical analysis, and as 2:1 by ^{13}C -n.m.r. studies. As the genera of *Pilophoron* and *Stereocaulon* are also members of the *Stereocaulaceae*, the present result seems quite reasonable. Accordingly, it may be concluded that the chemical taxonomy of lichens on the basis of their polysaccharide content is readily achieved by examining the ^{13}C -n.m.r. spectra thereof.

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REFERENCES

- 1 N. B. CHANDA, E. L. HIRST, AND D. J. MANNERS, *J. Chem. Soc.*, (1957) 1951–1958.
- 2 B. LINDBERG AND J. MCPHERSON, *Acta Chem. Scand.*, 8 (1954) 985–988.
- 3 T. TAKEDA, Y. NISHIKAWA, AND S. SHIBATA, *Chem. Pharm. Bull.*, 18 (1970) 1074–1075.
- 4 T. TAKEDA, M. FUNATSU, S. SHIBATA, AND F. FUKUOKA, *Chem. Pharm. Bull.*, 20 (1972) 2445–2449.
- 5 V. M. MIČOVIĆ, M. HRANISAVLJEVIĆ-JAKOVLJEVIĆ, AND J. MILJKOVIĆ-STOJANOVIĆ, *Carbohydr. Res.*, 10 (1969) 525–533.
- 6 Y. NISHIKAWA, T. TAKEDA, S. SHIBATA, AND F. FUKUOKA, *Chem. Pharm. Bull.*, 17 (1969) 1910–1916.
- 7 S. SHIBATA, in G. BENZ AND J. SANTESSON (Eds.), *Medicine and Natural Sciences: Chemistry in Botanical Classification*, Academic Press, New York, 1973, pp. 241–249.
- 8 S. SHIBATA, *J. Natl. Sci. Council. Sri Lanka*, 1 (1973) 183–188.
- 9 S. SHIBATA, Y. NISHIKAWA, M. TANAKA, F. FUKUOKA, AND M. NAKANISHI, *Z. Krebsforsch.*, 71 (1968) 102–104.
- 10 S. SHIBATA, Y. NISHIKAWA, C. F. MEI, F. FUKUOKA, AND M. NAKANISHI, *Gann*, 59 (1968) 159–161.

- 11 Y. Y. MAEDA, J. HAMURO, Y. YAMADA, K. ISHIMURA, AND G. CHIHARA, *Ciba Found. Symp.*, 18 (1973) 259-286.
- 12 I. YOKOTA AND S. SHIBATA, *Chem. Pharm. Bull.*, 26 (1978) 2668-2670.
- 13 P. COLSON, H. J. JENNINGS, AND I. C. P. SMITH, *J. Am. Chem. Soc.*, 96 (1974) 8081-8087.
- 14 T. USUI, N. YAMAOKA, K. MATSUDA, K. TUZIMURA, H. SUGIYAMA, AND S. SETO, *J. Chem. Soc. Perkin Trans. 1*, (1973) 2425-2432.